

## Analysis of a Sulfonamide Library.

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### 1. Introduction

- Quality and safety requirements expected for new drug compounds confront analytical chemists to the necessity of developing new analytical methods capable of quick, highly-efficient separations for the characterization of all compounds and impurities.
- Until recently HPLC-MS has been preferentially used for this purpose. However SFC-MS appears more and more as a complementary technique for high throughput analysis.
- Because HPLC and SFC are complementary, more and more analytical departments are being equipped with both types of instrumentation. To avoid time-consuming double analysis, it is of interest to be able to determine which technique will be the most efficient, for a given sample, prior to analysis.

### 2. Objectives

- Kaliskan and co-workers have studied prediction of the retention of diverse drug-like compounds in HPLC using three molecular descriptors: total dipole moment,  $\mu$ ; electron excess charge of the most negatively charged atom,  $\delta_{min}$ ; and water-accessible molecular surface area,  $A_{water}$ .<sup>1</sup> In SFC, studies have been undertaken for restricted sets of structurally similar analytes.<sup>2</sup>
- The applicability of SFC to drug-like compounds and the possibility of applying Kaliskan's model to predict the retention behaviour of a given compound are studied herein by screening a library of 32 sulfonamides.

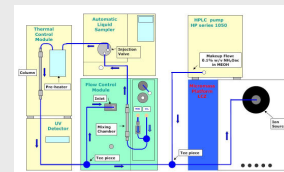
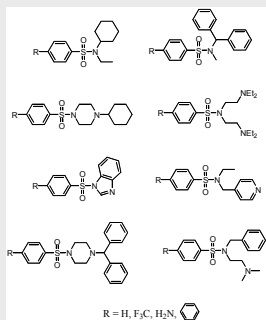
### 3. Instrumentation and Method

#### Instrumentation:

- Experiments undertaken on *SFC Berger MiniGram System* from Mettler Toledo.
- In addition to the UV detector, a *Mass Spectrometer Platform LCZ* is fitted to the system via a T-piece immediately after the UV detector outlet.
- In order to guarantee good ionisation of the analytes, a *makeup flow* is pumped into the system by a HPLC pump through another T-piece immediately before the MS inlet.

#### Method:

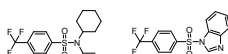
- "*Polycratic*" study: capacity ratios  $k$  measured for each compound at 10 different eluent compositions  $\phi$ .
- Over  $1 < k < 10$ , *log k vs.  $\phi$  relationship proved linear*.<sup>3</sup>  $\log k = \log k_0 + S \times \phi$ .
- Regression analysis*<sup>4</sup> performed to obtain values of the slope  $S$  and intercept  $\log k_0$ .
- Various molecular descriptors* (total dipole moment  $\mu$ , atomic formal charges and electron density surfaces) calculated using Spartan'02 software.
- Multiple regression analysis* performed to correlate  $S$ ,  $\log k_0$  and  $\phi_0 = -\log k_0/S$  with the calculated molecular descriptors.<sup>5</sup>



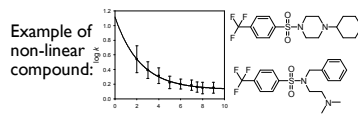
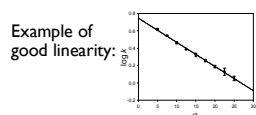
### 4. Results

#### Acquisition of chromatographic data and restriction of the test set

- Stationary phase:** test analytes were studied on packed **2-ethyl-pyridyl** column (4.6 x50mm, 60Å pores, 6 mm particle size).
- Mobile phase:** CO<sub>2</sub> was modified with methanol (MeOH) containing either 0.6mM of NH<sub>4</sub>OAc or 0.1% v/v of ethyl-dimethyl-amine (EDMA).
- Adjusting retention within  $0 < k < 1$ :**
  - retention of *late eluting compounds* could be adjusted by increasing the modifier proportion in the mobile phase.
  - three *early eluting analytes* were removed from the study due to lack of retention even at very low modifier concentration:



- Linearity of  $\log k = \log k_0 + S \times \phi$ :** good linearity was observed ( $R^2 > 0.98$ ) for all the analytes but two. Those two outliers were studied at concentrations of MeOH less than 10% v/v which is believed to explain the non-linearity.



- A few other studied analytes were not taken into account in the final calculations due to tailing or splitting peaks, examples are given below:



- All in all, 23 compounds fulfilled all requirements and were included in the final calculation with EDMA as additive and 21 with NH<sub>4</sub>OAc as additive.

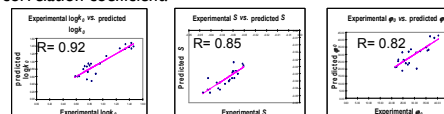
- Correlation of  $\log k_0$ ,  $S$  and  $\phi_0$  with molecular descriptors:** retention characteristics of the analytes were found to be correlated with calculated molecular descriptors: total dipole moment  $\mu$ , surface area  $A$  and atomic charge on the most negatively charged atom  $\delta_{min}$ .

$$\log k_0 = a \mu + b A - c \delta_{min} + d \quad S = e \mu + f A - g \delta_{min} + h$$

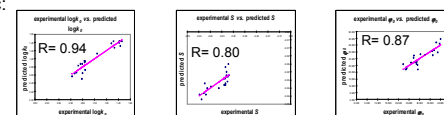
$$\phi_0 = i \mu + j A - l \delta_{min} + m$$

- Plots of experimental vs. predicted values:** these equations allow for the calculations of predicted values of the retention characteristics, that can be plotted vs. the experimental values with good correlation coefficient.

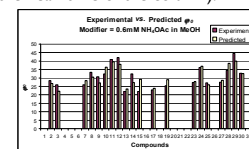
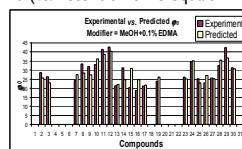
- MeOH + 0.1% v/v EDMA:



- MeOH + 0.6mM NH<sub>4</sub>OAc:



- Predicted values give good estimates of experimental data**, especially for  $\phi_0$ , which practically represents the proportion of modifier needed in the mobile phase to achieve  $k = 0$  (i.e. retention time equals 2 times the dead time of the column).



### 5. Conclusion and future work

- Polycratic retention studies** carried out for 32 sulfonamides on 2-ethyl-pyridyl stationary phase.
- When EDMA used as additive:** 23 compounds exhibited satisfactory retention and peak as well as linearity of the relationship  $\log k = \log k_0 + S \times \phi$   $R^2 > 0.98$ .
- When NH<sub>4</sub>OAc used as additive:** 21 compounds exhibited satisfactory retention and peak as well as linearity of the relationship  $\log k = \log k_0 + S \times \phi$   $R^2 > 0.98$ .
- With both additives, **retention characteristics  $\log k_0$ ,  $S$  and  $\phi_0$**  are correlated with molecular descriptors  $\mu$ ,  $A$  and  $\delta_{min}$ .
- Results to be considered with care**, since obtained on small set of structurally similar compounds.
- The study has to be extended to **higher concentration of additive** in the modifier, **other stationary phases** and **compounds of different structures**.
- Regression analysis using **other descriptors** has to be undertaken in an attempt to highlight better correlation.

### Acknowledgements



### References

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- G.A. Alvarez, W. Baumann *Zeitschrift Fur Naturforschung Section a-a Journal of Physical Sciences* **2005**, 60 (2005) 61-69.
- Schoenmakers, P. J.; Billiet, H. A. H.; De Galan, L. *Journal of Chromatography* **1979**, 185, 179-195.
- Regression analysis performed using SigmaPlot 9.0.
- Multiple regression analysis performed using Microsoft® Office Excel.