# CHEM3023: Spins, Atoms and Molecules

#### Lecture 5

# **The Hartree-Fock method**

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Learning outcomes

- Be able to use the variational principle in quantum calculations
- Be able to construct Fock operators for molecules
- Be able to explain the stages of a Hartree-Fock calculation



# So far:

- We wanted to construct the simplest possible electronic wavefunction, so we chose to use products of spin orbitals (one-electron wavefunctions)
- The requirement of antisymmetry forced us to use sums of products: these can be represented as a Slater determinant

$$\Psi_{SD}(\mathbf{x}_1, \mathbf{x}_2, \dots, \mathbf{x}_N) = \frac{1}{\sqrt{N!}} \begin{vmatrix} \chi_1(\mathbf{x}_1) & \chi_2(\mathbf{x}_1) & \dots & \chi_N(\mathbf{x}_1) \\ \chi_1(\mathbf{x}_2) & \chi_2(\mathbf{x}_2) & \dots & \chi_N(\mathbf{x}_2) \\ \vdots & \vdots & & \vdots \\ \chi_1(\mathbf{x}_N) & \chi_2(\mathbf{x}_N) & \dots & \chi_N(\mathbf{x}_N) \end{vmatrix}$$

• Spin orbitals have a spatial part and a spin part (either "up" or "down")

$$\begin{array}{l} \chi^{\uparrow}(\mathbf{x}) = \psi(\mathbf{r}) \alpha(\omega) \\ \text{Spin} \quad \text{Spatial Spin} \\ \text{orbital orbital function} \\ \chi^{\downarrow}(\mathbf{x}) = \psi(\mathbf{r}) \beta(\omega) \end{array}$$

• We now need a way to calculate the best possible spatial orbitals



# Finding the optimum orbitals

- How do we decide what are the "best" orbitals for a molecule?
- We need a guiding principle
- This is the variational principle: It can be applied to any approximate wavefunction.
- It forms the basis of many computational techniques
- To understand the variational principle we need the following two results for the (exact) eigenfunctions of any Hermitian operator:
  - 1) The are orthogonal (this is shown at the appendix)
  - 2) They form a complete set of functions (next slide)



#### The eigenfunctions of a Hermitian operator form a complete set

I.e. **Any function** (so, also any wavefunction) can be written exactly as a linear combination of the eigenfunctions of a Hermitian operator



• Proof is beyond this course

•We can understand this result by analogy with ordinary 3D space where any vector **r** can be expanded as a linear combination of the unit vectors along the x, y, z axes



# The variational principle

 $|\Phi
angle\,$  is an approximate ground state eigenfunction for the Hamiltonian  $\hat{H}$ 

As usual, its energy is given by 
$$E_0^{
m approx.}=rac{\langle\Phi|\hat{H}|\Phi
angle}{\langle\Phi|\Phi
angle}$$
 (1)

The exact wavefunctions are the eigenfunctions:

 $\hat{H}|\Psi_i\rangle = E_i|\Psi_i\rangle$ 

Remember that these are a complete set and so we can use them to expand our approximate wavefunction

$$|\Phi\rangle = \sum_{i} c_{i} |\Psi_{i}\rangle$$

Next, substitute this expansion into (1)



# The variational principle (continued)

The numerator becomes:

$$\langle \Phi | \hat{H} | \Phi \rangle = \langle \left( \sum_{i} c_{i} \Psi_{i} \right) | \hat{H} | \left( \sum_{j} c_{j} \Psi_{j} \right) \rangle$$

$$= \sum_{i} \sum_{j} c_{i}^{*} c_{j} \langle \Psi_{i} | \hat{H} | \Psi_{j} \rangle = \sum_{i} \sum_{j} c_{i}^{*} c_{j} \langle \Psi_{i} | E_{j} | \Psi_{j} \rangle$$

$$= \sum_{i} \sum_{j} c_{i}^{*} c_{j} E_{j} \langle \Psi_{i} | \Psi_{j} \rangle = \sum_{i} \sum_{j} c_{i}^{*} c_{j} E_{j} \delta_{ij} = \sum_{i} |c_{i}|^{2} E_{i}$$

$$(2)$$

In a similar way, the denominator becomes:

$$\langle \Phi | \Phi 
angle = \sum_i |c_i|^2$$
 (3)



### The variational principle (continued)

By combining (2) and (3) we obtain a relation for  $E_0^{
m approx.}$  :

$$E_0^{\text{approx.}} = \frac{\langle \Phi | \hat{H} | \Phi \rangle}{\langle \Phi | \Phi \rangle} = \frac{\sum_i E_i |c_i|^2}{\sum_j |c_j|^2} \ge \frac{\sum_i E_0 |c_i|^2}{\sum_j |c_j|^2} = \frac{E_0 \sum_i |c_i|^2}{\sum_j |c_j|^2} = E_0$$
  
by definition, the ground state energy is the lowest  
 $\dots E_3 \ge E_2 \ge E_1 \ge E_0$ 

Therefore:  $E_0^{\text{approx.}} \ge E_0$  for all approximate wavefunctions

- The exact ground state energy is lower than any approximation
- Therefore any change to an approximate wavefunction that lowers the energy is an improvement



### **The Hartree-Fock equations**

Starting from the energy of a Slater determinant wavefunction:

$$E_{SD} = 2\sum_{i=1}^{N_{elec}/2} h_i + \sum_{i=1}^{N_{elec}/2} \sum_{j=1}^{N_{elec}/2} (2J_{ij} - K_{ij})$$

It can be shown (see e.g. Szabo & Ostlund – or ask me outside the lectures) that the energy of the Slater determinant is minimised when the molecular orbitals are solutions of the following eigenvalue equation:

$$\hat{f}\psi_i(\mathbf{r}) = \varepsilon_i\psi_i(\mathbf{r})$$

This is the Fock operator



# The Fock operator

 $\hat{f} = \hat{h} + \sum_{i=1}^{N_{elec}/2} \left( 2\hat{J}_i - \hat{K}_i \right)$ 

i=1

$$\hat{f}\psi_i(\mathbf{r}) = \varepsilon_i\psi_i(\mathbf{r})$$

Is built from three operators:

- Core Hamiltonian
- •Coulomb
- $\hat{J}_i(\mathbf{r}_1)\psi_k(\mathbf{r}_1) = \left[\int \psi_i^*(\mathbf{r}_2)rac{1}{r_{12}}\psi_i(\mathbf{r}_2)d\mathbf{r}_2
  ight]\psi_k(\mathbf{r}_1)$  Exchange
  - Coulomb operator

**Core Hamiltonian** 

operator

$$\hat{K}_{i}(\mathbf{r}_{1})\psi_{k}(\mathbf{r}_{1}) = \left[\int \psi_{i}^{*}(\mathbf{r}_{2})\frac{1}{r_{12}}\psi_{k}(\mathbf{r}_{2})d\mathbf{r}_{2}\right]$$

 $\hat{h} = -\frac{1}{2}\nabla^2 + \sum_{i=1}^{N_{nuc}} \frac{-Z_I}{|\mathbf{r} - \mathbf{R}_I|}$ 



#### Fock operator example with 4 electrons





#### **Example of the Hartree-Fock equations**

- Let's write down the Hartree-Fock eigenvalue equation for a molecule with 6 electrons, e.g. BeH<sub>2</sub>
- Observe the structure of this equation. Can you suggest a way of solving it?



 $\psi_5$ 



# The Fock operator (continued)

- Is Hermitian (try to show this!)
- Its eigenfunctions are the optimal molecular orbitals we seek
- Its eigenvalues are the orbital energies
- It consists of three parts:
  - Core Hamiltonian operator: kinetic energy of an electron and its electrostatic attraction to all the nuclei
  - Coulomb operator: Electrostatic repulsion of an electron from all other electrons
  - Exchange operator: A purely quantum effect (no classical analogue), consequence of the antisymmetry of the electronic wavefunction



#### The Self-Consistent Field (SCF) procedure

- The Fock operator depends on the molecular orbitals which we seek
- Therefore an iterative procedure needs to be used to find these orbitals
- This procedure is often called a Self-Consistent Field (SCF) calculation





# **Summary / Reading assignment**

- Variational principle (Cramer, pages 108-110)
- Hartee-Fock method (Cramer, pages 126-129)



#### Appendix: The eigenfunctions of a Hermitian operator are orthogonal

Assume A is a Hermitian operator. Let's pick two of its non-degenerate eigenfunctions:

•Therefore non-degenerate eigenfunctions are orthogonal

•Degenerate eigenfunctions can always be chosen to be orthogonal

